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(54) **Epoxy-urethane imine primer**

(57) Novel coating compositions are disclosed. In particular, novel coating compositions are disclosed that comprise the reaction product of epoxy, imine, isocyanate and optionally an active hydrogen compound and/or or active hydrogen containing compounds. Wherein those reaction products are called polyuroxy compounds and wherein hydroxyl containing compounds are the preferred active hydrogen containing compounds. The disclosed compounds can be used as refinish and or sealer compositions.

**Description****FIELD OF THE INVENTION**

5 [0001] The present invention is directed to novel coating compositions. In particular, the present invention is directed to polyureoxy coating compositions that comprise an epoxy, imine (preferably ketimine), isocyanate and optionally an active hydrogen compound or an active hydrogen containing compound, where for the purpose of this invention the term polyureoxy describes the reaction product of polyurethanes, polyurea and epoxy amine. The compositions have been found to be excellent refinish or sealer compositions, are tintable, and may be applied, wet on wet, utilizing a base-coat/clearcoat or single stage topcoat over them.

**BACKGROUND OF THE INVENTION**

15 [0002] With increasing Volatile Organic Content (VOC) restrictions, it is difficult to get a balance of pot life and dry time for coatings useful in refinish paint compositions. The main problems have been that the hydroxyl functional polyols are generally high molecular weight resins, which are viscous materials and require a lot of organic solvents in order to reduce to sprayable viscosities. It is also difficult to get the VOC low enough to be compliant with environmental restrictions. By using reactive diluents in combination with the polyol and isocyanates, lower VOC can be achieved. An excellent example of this type of coating is exemplified in U.S. Patent No. 5,14,086 and is assigned to BASF Corporation.

20 [0003] In the prior art, the use of diketimines, dialdimines, and hindered diamines as reactive diluents, helps to lower the viscosity and increase the solids, thus enabling coating compositions to reach required VOC limits. However, as more stringent VOC limits are being set, it has become necessary to try and achieve even lower VOC levels. Low molecular weight hydroxyl functional polyols are difficult to use in order to reach these VOC limits. In addition, these polyols react very fast and therefore it is difficult to achieve a sufficient pot life.

25 [0004] An approach to a coating composition has been to mix epoxy reactants and/or epoxy/amine reactants with urethane and/or isocyanate reactants but because amines react exceedingly faster with epoxy compounds than with the urethane or isocyanate compounds they leave epoxy or hydroxyl groups which do not react under ambient conditions with isocyanate.

30 [0005] We have now found that polyureoxy coating compositions overcome the prior art disadvantages and result in low VOC coatings that are especially useful in refinish compositions.

[0006] The present invention is directed to novel coating compositions containing polyureoxy compounds. In particular, the present invention is directed to coating compositions that comprise the reaction product of epoxy, imine, isocyanate and/or active hydrogen compound(s) and/or active hydrogen containing compound(s).

**SUMMARY OF THE INVENTION**

35 [0007] The present invention relates to the use of polyureas and/or polyureoxy compositions in coating systems especially coatings suitable for the production of paint for the automotive industry. The coatings can be cured at ambient conditions or can be cured by baking and are therefore useful in refinish paint compositions. The coating composition optionally contains pigments and other well known paint additives such as fillers, rheology control agents, dispersing agents, solvents, etc.

[0008] The coating composition contains.

- 45 a) at least one epoxy functional compound,
- b) an imine functional compound.
- c) at least one isocyanate functional compound
- d) and optionally an active hydrogen compound and/or an active hydrogen containing compound.

50 [0009] The coating composition can be low in volatile organic content (VOC), but the invention is not limited to only low VOC compositions. The invention also relates to the coating composition's reaction behavior which is moisture dependent, though moisture is not essential for the reaction. The invention also relates to optionally using catalysts to accelerate the reaction rate of the active hydrogen compound of the imine functional compound, secondary amine and aldimine compounds with the isocyanate compounds. The catalysts used in this invention are aromatic or aliphatic carboxylic acids, inorganic acids such as sulfonic acids, organic tin based compounds or arylsulfonic acids as described in the cited prior art. The use of moisture scavengers in order to control the reaction rates in humid conditions is optional.

DETAILED DESCRIPTION OF THE INVENTION

[0010] The present invention relates to the use of polyureas and/or polyureoxy compositions in coating systems especially coatings suitable for the production of paint for the automotive industry. The coatings can be cured at ambient conditions or by baking and are therefore useful in refinish paint compositions. The coating composition optionally contains pigments and other well known paint additives such as fillers, rheology control agents, dispersing agents, solvents, etc.

[0011] The coating composition contains..

- a) at least one epoxy functional compound,
- b) an imine functional compound,
- c) at least one isocyanate functional compound
- d) and optionally an active hydrogen compound and/or an active hydrogen containing compound.

[0012] In the coating compositions according to the invention the polyureoxy parts are created in situ during the curing process:

1.) The polyurethane is obtained by reaction of isocyanate groups with "activated" epoxy functional compounds (meaning per addition to the epoxy group formed hydroxyl groups or the like) and/or active hydrogen (containing) compounds.

2.) The polyurea is obtained by reaction of the imine, either as an enamine or after hydrolisation as an amine, with isocyanate groups.

3.) The epoxy amine is obtained from the epoxy compound and the imine (as enamine or amine).

[0013] This invention also describes a method for reducing the volatile organic content (VOC) of a paint composition, by increasing the solids in a paint composition without adversely affecting the application, durability or performance of the composition.

[0014] Useful epoxy compounds are those which react with amine or imine functional compounds under ambient or bake conditions. Optionally these epoxides may be reactive with other active hydrogen containing compounds.

[0015] Useful isocyanates are di- or polyisocyanates which are aliphatic, cycloaliphatic, or aromatic. Such isocyanates include hexamethylene diisocyanate, 1-methyl-2, 4-diisocyanatocyclo-hexane, isophorone diisocyanate, 4,4-diisocyanatodicyclohexyl methane, toluene-2-4-diisocyanate, o-, m-, and p-xylene diisocyanate, 1-5-naphthalene diisocyanate or mixtures thereof. It is also possible to use polyisocyanates with isocyanurates, allophanate or uretdione structures. The polyisocyanates may optionally be reacted with a deficit of polyhydroxy compounds, such as water, glycols, polyalkylene glycols, neopentyl glycols, glycerol, trimethylol propane, hexane triol, or alkyd resins, before being used.

[0016] This invention also relates to a process of formulating coatings by using aldimine and ketimine. Typically, the aldimine does not completely hydrolyze into amine and aldehyde, but probably reacts with the isocyanate as the enamine.

[0017] Useful hydroxyl containing compounds are acrylic polyols, polyester polyols and those known to the prior art.

[0018] The composition may also contain pigments. These pigments can be introduced by first forming a mill base with the secondary amine functional compound and/or epoxy functional compound or optionally with the aldimine by conventional techniques, such as sand grinding, ball milling, attritor grinding, or two roll milling to disperse the pigments. The mill base is blended with the film forming constituents as shown in the examples which follow.

[0019] The invention also relates to optionally using catalysts to accelerate the reaction rate of the epoxy and imine compounds with isocyanate compounds. The catalysts to be used in this invention may be organic acids, such as aromatic or aliphatic carboxylic acids as well as arylsulfonic acids, inorganic acids such as sulfonic acids, or organic tin based compounds.

[0020] The use of moisture scavenging materials is optional in this application and may be used in order to control the water which is released from the pigments, especially in grinding and to control the reaction rates in humid conditions. The removal of this water in the mill base helps to control the reaction rates of the coating composition and allow for the desired properties of the coating film.

[0021] Coating compositions described by the present invention find utility in application of ambient film forming and curing such as automotive refinish coatings. It is also suggested that the present invention applied to coatings to be force dried or baked to accelerate the coating curing process. Forced dry conditions range from 38°C (100 degrees Fahrenheit) to over 163°C (325 degrees Fahrenheit). The coating cure process for the present invention may also be accelerated by the utilization of radiant heating or Infra Red emitting devices known to the art.

[0022] The following examples are intended to illustrate the invention.

[0023] The following epoxy-isocyanate-imine (polyureoxy) primer systems were prepared and achieved a 1.41 kg/s (3.1

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lbs.) VOC or less and exhibited excellent adhesion, humidity resistance, salt spray resistance and chip resistance. Typical vehicle compositions are listed in the following examples (1-3):

The specific use as a tintable sealer is illustrated in the examples (4, 5), which use actual BASF automotive refinish products to prepare the tinted polyureoxy sealer. Two examples of a tinted sealer are provided. These examples are the prime embodiment of polyureoxy as a tintable sealer. They are especially useful as a sealer under a poor hiding basecoat/clearcoat topcoat, or poor hiding single-stage topcoat. The examples illustrated here utilize a BASF single-stage topcoat, HS SOLO.

### EXAMPLES

#### Example 1

[0024]

	[%] by weight	[%] by weight
Epoxy resin (Dow DER 663U)	21.8	34.3
Butyl benzyl phthalate (Monsato Santicizer 160)	5.7	9.3
Ketimine resin (Akzo Setalux K (7002 bX-55)	9.7	14.4
Hydroxyl acrylic (BASF Acrylic Resin)	22.9	---
Isocyanurate trimer (Rhône-Poulenc Tolonate HDT)	39.9	42.0
Total	100	100

#### Example 2

[0025]

	[%] by weight
Epoxy resin (SHELL Epon 1001-X-75)	16.8
Epoxy ester resin (BASF Epoxy Ester)	14.5
Ketimine (BAYER LS-2965)	33.1
Isocyanate trimer (Rhône-Poulenc Tolonate HDT-LV)	9.6
Isocyanate trimer (BAYER N-3400)	26.0
Total	100

#### Example 3

[0026]

	[%] by weight
Epoxy resin (Dow DER 663U)	34.3
Butyl benzyl phthalate (Monsato Santicizer 160)	9.3
Aldimine (BAYER XP-7076)	14.4

(continued)

	[%] by weight
Isocyanate trimer (Rhône-Poulenc Tolonate HDT)	42.0
Total	100

Examples 4, 5Tintable polyureoxy sealers

[0027]

	Example 4 [%] by volume	Example 5 [%] by volume
BASF white epoxy primer EP-689	33	33
BASF single-stage red tint base HS SOLO HS-82	17	--
BASF single-stage yellow tint base HS SOLO HS-62	--	17
BASF ketimine polyureoxy activator - UA-97	33	33
BASF isocyanate hardener - DH-46	17	17
Total	100	100

[0028] The red polyureoxy sealer of example 4 is indicative of the formula cited as example 6. This sealer was top-coated with BASF HS SOLO topcoat red. Adequate topcoat color development was achieved with two coats of the single stage topcoat, as measured by computer measurement of Delta E from the standard. Normally, if the H86404 red single-stage topcoat were applied over an untinted epoxy sealer, it would require 3-4 coats to achieve an adequate color match to the standard. Therefore, the use of the polyureoxy technology as a tintable sealer saves time and money. It should be mentioned, that only one coat of polyureoxy sealer was applied.

[0029] The yellow polyureoxy sealer of example 5 is indicative of the formula cited as example 1. This sealer was top-coated after 15-20 minutes flash off time, with BASF HS SOLO single-stage topcoat yellow. One coat of polyureoxy tintable sealer was applied, followed by 2 coats of H80755 single-stage yellow. At this point, adequate color development was achieved, as measured by computer measurement of Delta E from the standard. For comparison purposes, the same H80755 single-stage yellow topcoat was applied over a standard untinted epoxy primer, which was gray. In this case, it took 4 coats of the H80755 yellow to achieve adequate color match, to the standard.

[0030] In summary, the polyureoxy technology allows the combination of epoxy technology with hydrofunctional resin systems, such as described in U.S. 5,214,086. This allows the epoxy component to be tinted any color, preferably to match closely to that of the topcoat, so that time and money may be saved, because fewer consecutive coating processes are necessary. The role of the ketimine activator and isocyanate hardener, are especially important to the polyureoxy technology, in that they allow the entire system to become an integrally crosslinked network.

**Claims**

1. A polyureoxy coating composition containing:

- a) at least one epoxy functional compound,
- b) an imine functional compound,
- c) at least one isocyanate functional compound
- d) and optionally an active hydrogen compound and/or an active hydrogen containing compound.

2. A composition according to claim 1 which is cured at ambient temperatures.

3. A composition according to claim 1 which is baked to cure.

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4. A composition according to claim 1 that additionally contains at least one pigment.
5. A composition according to claim 1 wherein the isocyanate functional compound is chosen from aliphatic, aromatic or cycloaliphatic mono, or diisocyanates, triisocyanates, uretdiones, biuret or isocyanate functional polymers.
6. A composition according to claim 5 wherein the isocyanate functional compound is a dimer or trimer of hexamethylene diisocyanate or a blend of both.

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European Patent  
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# EUROPEAN SEARCH REPORT

Application Number  
EP 98 12 3301

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
X	US 3 993 708 A (BRINKMANN BERND ET AL) 23 November 1976 * column 1, line 8 - column 4, line 68 * * examples 2,4,5 * ----	1,2,5	C08G18/58 C08G18/64 C09D5/00
X	US 5 034 495 A (HANSEN ACHIM ET AL) 23 July 1991 * column 1, line 58 - column 3, line 31 * * examples 1-3,6 * * claims 1,10,11 * ----	1,2,5	
X	US 5 071 951 A (ULRICH GRUNOKE ET AL) 10 December 1991 * column 1, line 61 - column 2, line 64 * * examples 1,2,5,11,12 * * claims 1,11,12 * ----	1,3,5	
X	US 5 227 414 A (ERNST G DALE ET AL) 13 July 1993 * column 2, line 46 - column 5, line 19 * * examples 2-5; tables 1-3 * ----	1,2,5	
A	US 4 557 814 A (SCHUPP EBERHARD ET AL) 10 December 1985 * column 2, line 47 - column 5, line 19 * * examples 1-5; tables 2,3 * -----	1,2,5	
The present search report has been drawn up for all claims			
Place of search <b>THE HAGUE</b>		Date of completion of the search <b>22 February 1999</b>	Examiner <b>Neugebauer, U</b>
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ..... &amp; : member of the same patent family, corresponding document</p>			

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ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.

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This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.  
The members are as contained in the European Patent Office EDP file on  
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22-02-1999

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 3993708 A	23-11-1976	DE 2418041 A	06-11-1975
		AT 347691 B	10-01-1979
		AT 278575 A	15-05-1978
		CH 619474 A	30-09-1980
		DE 2462791 C	23-12-1982
		FI 750700 A, B,	14-10-1975
		FR 2267335 A	07-11-1975
		GB 1496432 A	30-12-1977
		JP 1306902 C	13-03-1986
		JP 50139200 A	06-11-1975
		JP 60025446 B	18-06-1985
		JP 1483023 C	27-02-1989
		JP 57123215 A	31-07-1982
		JP 63032087 B	28-06-1988
		NL 7504390 A, B,	15-10-1975
		SE 415771 B	27-10-1980
		SE 7504124 A	19-12-1975
US 5034495 A	23-07-1991	DE 3919128 A	13-12-1990
		DE 59005586 D	09-06-1994
		EP 0402579 A	19-12-1990
		ES 2018753 T	16-07-1994
		GR 90300187 T	10-10-1991
US 5071951 A	10-12-1991	DE 3919547 A	20-12-1990
		EP 0402582 A	19-12-1990
		GR 90300185 T	10-10-1991
US 5227414 A	13-07-1993	NONE	
US 4557814 A	10-12-1985	DE 3325061 A	24-01-1985
		AT 27608 T	15-06-1987
		BR 8403432 A	25-06-1985
		CA 1232099 A	26-01-1988
		EP 0134983 A	27-03-1985
		JP 1756377 C	23-04-1993
		JP 4049843 B	12-08-1992
		JP 60063223 A	11-04-1985
		PT 78887 B	02-06-1986

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82